

# Simple Concentration Analysis of Flavor Components in Roasted Laver - Using MonoTrap Collector Tools

MonoTrap RGC18TD, a simple concentration tool was used to simply concentrate the flavor components in roasted laver. These were then introduced and measured by thermal desorption. Easy-to-use, high-sensitivity analysis was achieved. Dimethyl sulfide and  $\beta$ -ionone, which are the fragrance components of smelling of the sea were both detected.

## Pretreatment procedure

Roasted laver

Cut one piece with scissors.

Collection (HS)  
One MonoTrap RGC18TD

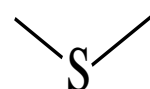
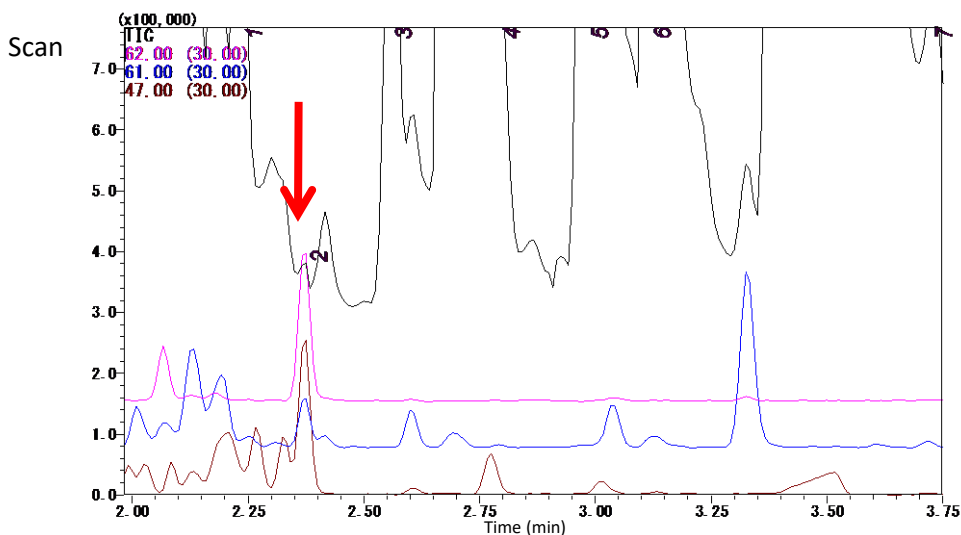
40 °C for 3 h

TD-GC-MS



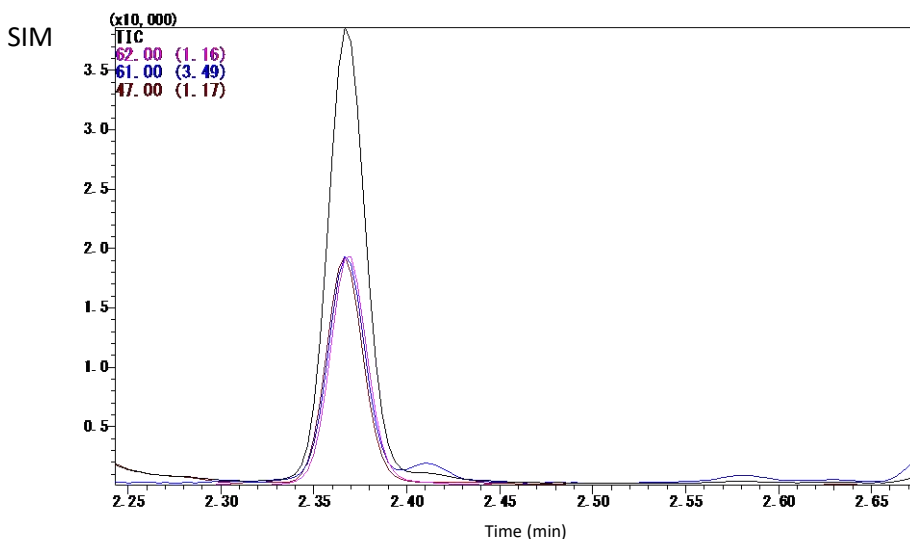
### Conditions

<b>System</b>	: GC-MS-Thermal Desorption
<b>Column</b>	: InertCap Pure-WAX 0.25 mm I.D. × 30 m df = 0.25 $\mu$ m
<b>Col.Temp.</b>	: 40 °C (5 min) - 6 °C/min - 250 °C
<b>Carrier Gas</b>	: He 1 mL/min (constant flow)
<b>Desorb Temp.</b>	: 200 °C
<b>Time</b>	: 5 min
<b>Flow</b>	: 5 mL/min
<b>Split</b>	: Splitless
<b>Cryo Trapping</b>	: -150 °C
<b>Injection Temp.</b>	: 250 °C
<b>Detection</b>	: MS Scan (m/z = 28.5 - 600) SIM (m/z : 62,61,47 for Dimethyl sulfide)

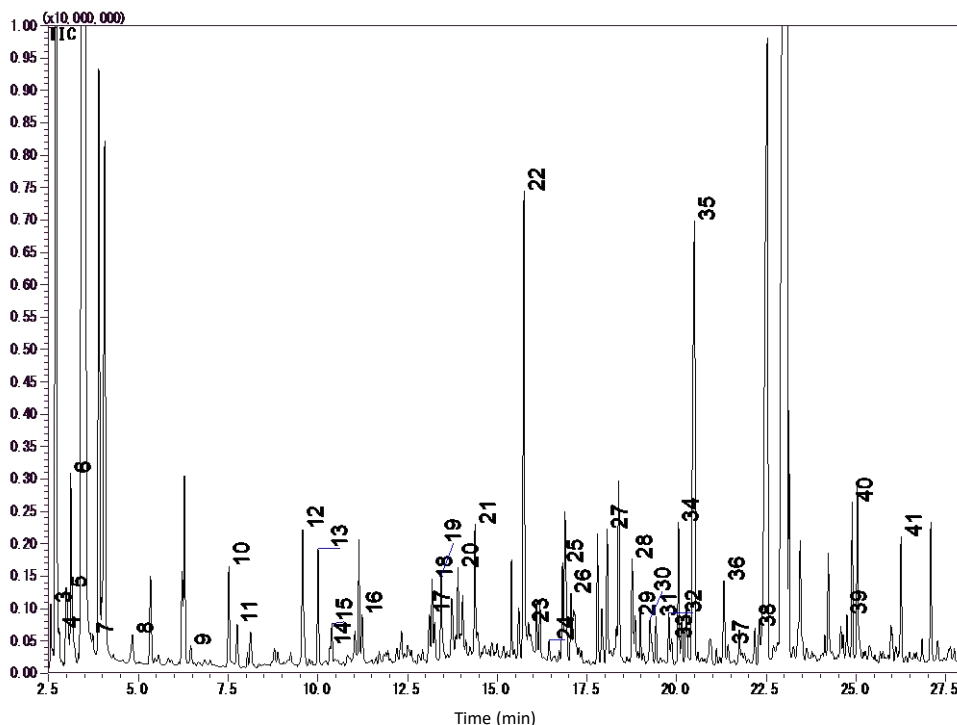


Dimethyl sulfide  
m/z = 62,61,47

Structures created using Chemistry 4-D Draw  
Which is provided by ChemInnovation Software, Inc.



For detecting Dimethyl sulfide in SIMs  
I confirmed.



- |                                    |   |
|------------------------------------|---|
| 1. Acetaldehyde                    | 22. Nonanal                             |
| 2. <b>Dimethyl sulfide</b>         | 23. Trimethylpyrazine                   |
| 3. Propanal                        | 24. 2-Octenal                           |
| 4. Methyl acetate                  | 25. <b>2-Ethyl-3,6-dimethylpyrazine</b> |
| 5. Trimethylamine                  | 26. 1-Octen-3-ol                        |
| 6. Trimethylamine                  | 27. Decanal                             |
| 7. <b>Isovaleraldehyde</b>         | 28. 2-Nonenal                           |
| 8. Pentanal                        | 29. 3-Caranol                           |
| 9. 1-Propanol                      | 30. Dimethyl Sulfoxide                  |
| 10. Hexanal                        | 31. 3,5-Octadien-2-one                  |
| 11. <b>2-Methyl-2-butenal</b>      | 32. <b>2,6-Nonadienal</b>               |
| 12. <b>2-Ethyl-trans-2-butenal</b> | 33. Propylene Glycol                    |
| 13. <b>1-Penten-3-ol</b>           | 34. 2,6-Dimethylcyclohexanol            |
| 14. 2-Heptanone                    | 35. <b>Butyrolactone</b>                |
| 15. Heptanal                       | 36. <b>n-Methyl-2-pyrrolidinone</b>     |
| 16. 2-Hexenal                      | 37. 2(5H)-Furanone, 5-methyl-           |
| 17. Acetoin                        | 38. 2-Dodecanone                        |
| 18. Octanal                        | 39. $\alpha$ -Ionone                    |
| 19. Acetol                         | 40. <i>Trans</i> -Geranylacetone        |
| 20. <b>2,5-Dimethylpyrazine</b>    | 41. <b><math>\beta</math>-Ionone</b>    |
| 21. 6-Methyl-5-heptene-2-one       |   |

\* For the qualitative analysis of standard samples Not performed. This results in a library search.

Red letters: flavor components of burning

(Reference: Summary of the 52nd Conference on Fragrances, Terpenes and Essential Oil Chemistry)

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